



Avogadro constant,  $N_A = 6.02 \times 10^{23} \text{ mol}^{-1}$   
 Ideal gas constant,  $R = 8.314 \text{ J K}^{-1} \text{ mol}^{-1}$   
 Volume 1 mole ideal gas at 100 kPa & 0°C (273.15 K),  $V_m = 22.71 \text{ L}$   
 Volume 1 mole ideal gas at 100 kPa & 25°C (298.15 K),  $V_m = 24.79 \text{ L}$   
 Ion product for water at 25°C (298.15 K),  $K_w = 1 \times 10^{-14}$   
 Specific heat capacity of water,  $c_g = 4.18 \times 10^3 \text{ J kg}^{-1} \text{ K}^{-1}$   
 Faraday constant,  $F = 96\,500 \text{ C mol}^{-1}$

JANUARY 2019

Mon	Tue	Wed	Thu	Fri	Sat	Sun
	1	2	3	4	<u>5</u>	<u>6</u>
7	8	9	10	11	<u>12</u>	<u>13</u>
14	15	16	17	18	<u>19</u>	<u>20</u>
21	22	23	24	25	<u>26</u>	<u>27</u>
28	29	30	31			

Assignments:

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Tests/Exams:

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Avogadro constant,  $N_A = 6.02 \times 10^{23} \text{ mol}^{-1}$   
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## FEBRUARY 2019

Mon	Tue	Wed	Thu	Fri	Sat	Sun
				1	<u>2</u>	<u>3</u>
4	5	6	7	8	<u>9</u>	<u>10</u>
11	12	13	14	15	<u>16</u>	<u>17</u>
18	19	20	21	22	<u>23</u>	<u>24</u>
25	26	27	28			

### Assignments:

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### Tests/Exams:

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MARCH 2019

Mon	Tue	Wed	Thu	Fri	Sat	Sun
				1	<u>2</u>	<u>3</u>
4	5	6	7	8	<u>9</u>	<u>10</u>
11	12	13	14	15	<u>16</u>	<u>17</u>
18	19	20	21	22	<u>23</u>	<u>24</u>
25	26	27	28	29	<u>30</u>	<u>31</u>

Assignments:

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Tests/Exams:

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Avogadro constant,

Ideal gas constant,

Volume 1 mole ideal gas at 100 kPa & 0°C (273.15 K),  $V_m = 22.71$  L

Volume 1 mole ideal gas at 100 kPa & 25°C (298.15 K),  $V_m = 24.79$  L

Ion product for water at 25°C (298.15 K),  $K_w = 1 \times 10^{-14}$

Specific heat capacity of water,

Faraday constant,

$$N_A = 6.02 \times 10^{23} \text{ mol}^{-1}$$

$$R = 8.314 \text{ J K}^{-1} \text{ mol}^{-1}$$

$$V_m = 22.71 \text{ L}$$

$$V_m = 24.79 \text{ L}$$

$$K_w = 1 \times 10^{-14}$$

$$c_g = 4.18 \times 10^3 \text{ J kg}^{-1} \text{ K}^{-1}$$

$$F = 96\,500 \text{ C mol}^{-1}$$

APRIL 2019

Mon	Tue	Wed	Thu	Fri	Sat	Sun
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30					

Assignments:

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Tests/Exams:

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$$V_m = 22.71 \text{ L}$$

$$V_m = 24.79 \text{ L}$$

$$K_w = 1 \times 10^{-14}$$

$$c_g = 4.18 \times 10^3 \text{ J kg}^{-1} \text{ K}^{-1}$$

$$F = 96\,500 \text{ C mol}^{-1}$$

MAY 2019

Mon	Tue	Wed	Thu	Fri	Sat	Sun
		1	2	3	<u>4</u>	<u>5</u>
6	7	8	9	10	<u>11</u>	<u>12</u>
13	14	15	16	17	<u>18</u>	<u>19</u>
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$$V_m = 24.79 \text{ L}$$

$$K_w = 1 \times 10^{-14}$$

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JUNE 2019

Mon	Tue	Wed	Thu	Fri	Sat	Sun
					<u>1</u>	<u>2</u>
3	4	5	6	7	<u>8</u>	<u>9</u>
10	11	12	13	14	<u>15</u>	<u>16</u>
17	18	19	20	21	<u>22</u>	<u>23</u>
24	25	26	27	28	<u>29</u>	<u>30</u>

Assignments:

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Tests/Exams:

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Mon	Tue	Wed	Thu	Fri	Sat	Sun
1	2	3	4	5	<u>6</u>	<u>7</u>
8	9	10	11	12	<u>13</u>	<u>14</u>
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Assignments:

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Tests/Exams:

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AUGUST 2019

Mon	Tue	Wed	Thu	Fri	Sat	Sun
			1	2	<u>3</u>	<u>4</u>
5	6	7	8	9	<u>10</u>	<u>11</u>
12	13	14	15	16	<u>17</u>	<u>18</u>
19	20	21	22	23	<u>24</u>	<u>25</u>
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SEPTEMBER 2019

Mon	Tue	Wed	Thu	Fri	Sat	Sun
						<u>1</u>
2	3	4	5	6	<u>7</u>	<u>8</u>
9	10	11	12	13	<u>14</u>	<u>15</u>
16	17	18	19	20	<u>21</u>	<u>22</u>
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OCTOBER 2019

Mon	Tue	Wed	Thu	Fri	<u>Sat</u>	<u>Sun</u>
	1	2	3	4	<u>5</u>	<u>6</u>
7	8	9	10	11	<u>12</u>	<u>13</u>
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				1	<u>2</u>	<u>3</u>
4	5	6	7	8	<u>9</u>	<u>10</u>
11	12	13	14	15	<u>16</u>	<u>17</u>
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## DECEMBER 2019

Mon	Tue	Wed	Thu	Fri	Sat	Sun
						<u>1</u>
2	3	4	5	6	<u>7</u>	<u>8</u>
9	10	11	12	13	<u>14</u>	<u>15</u>
16	17	18	19	20	<u>21</u>	<u>22</u>
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